# - HUMAN LENS CHEMISTRY -

### UV FILTERS AND AGE-RELATED NUCLEAR CATARACT

# A thesis submitted in partial fulfillment of the requirements for the award of the degree of

### DOCTOR OF PHILOSOPHY

### from

# MACQUARIE UNIVERSITY

by

# JASMINKA MIZDRAK B.Sc (Hons)

Department of Chemistry and Biomolecular Sciences

July, 2007

# TABLE OF CONTENTS

TAB	BLE OF C	CONTENTS	III
LIST	r of fig	URES	XI
LIST	r of sch	IEMES	XIX
LIST	Г OF TAI	BLES	XXIII
ABS	TRACT		XXV
THE	SIS DEC	LARATION	XXIX
		EDGMENTS	
		DNS	
		CE PRESENTATIONS	
		BREVIATIONS	
CHA	APTER 1		1
INTI	RODUCT	ION	1
1.1	General	I Introduction	1
1.2	The hur	nan eye	
1.3	The hur	nan lens	
1.4	Lens co	omposition	5
	1.4.1	Lens proteins	
	1.4.2	Lens antioxidants	
		1.4.2.1 Glutathione (GSH)	
		1.4.2.2 Ascorbic acid	7
	1.4.3	Lens UV filters	
1.5	Normal	human lens aging	
	1.5.1	Colouration and fluorescence in human lenses	
	1.5.2	Lens barrier formation	
	1.5.3	Decrease in antioxidants	
	1.5.4	Post-translational modifications of proteins	
		1.5.4.1 Neutral or subtractive modifications	
		1.5.4.1.1 Truncation, racemisation and deamidation	
		1.5.4.2 Additive modifications	
		1.5.4.2.1 Carbamylation, acetylation, phosphorylation and methylation	
		1.5.4.2.2 Glycation	
		1.5.4.3 UV filter modifications	
1.6	Catarac	t – Age-Related Nuclear cataract	
	1.6.1	Lens changes upon ARN cataract development	
	1.6.2	Possible factors in cataract formation	
		1.6.2.1 UV light and ARN cataract	
		1.6.2.2 Lens photosensitisers	
1.7	Aims		

	1.7.1	General aims	25		
	1.7.2	Specific aims	25		
CHA	PTER 2.		27		
A CO	NVENIEN	IT SYNTHESIS OF 30HKG	27		
2.1	Introducti	ion	27		
2.2	Results a	nd Discussion	29		
	2.2.1	Synthesis of 3-hydroxy-2-nitroacetophenone (9)	29		
	2.2.2	Synthesis of 4-(3-hydroxy-2-nitrophenyl)-4-oxobut-2-enoic acid (7)			
		2.2.2.1 Method 1 and 2	33		
		2.2.2.2 Method 3	34		
	2.2.3	Synthesis of methyl 4-(3-hydroxy-2-nitrophenyl)-4-oxobut-2-enoate (10)	35		
		2.2.3.1 Method 1 and 2	36		
		2.2.3.2 Method 3	38		
		2.2.3.3 Method 4	39		
	2.2.4	Synthesis of ethyl 4-(3-hydroxy-2-nitrophenyl)-4-oxobut-2-enoate (17)	40		
		2.2.4.1 Method 1	41		
		2.2.4.2 Method 2	41		
		2.2.4.3 Method 3	42		
		2.2.4.4 Method 4	43		
	2.2.5	Synthesis of methyl (11) and ethyl (18) 4-(2-nitro-3-((2,3,4,5-tetra-O-acetyl-β-D-			
		glucopyranosyl)oxyphenyl)-4-oxobut-2-enoate	44		
		2.2.5.1 Reaction trials	45		
		2.2.5.1.1 Reactions 1-3	45		
		2.2.5.1.2 Reactions 4-9	46		
		2.2.5.2   Method 1 (reaction 6 on a larger scale)	48		
		2.2.5.3Method 2 (reaction 9 on a larger scale)			
		2.2.5.4 Method 3			
	2.2.6	Synthesis of 3-hydroxykynurenine- <i>O</i> -β-D-glucoside (3OHKG)	50		
	2.2.7	Stability of 3OHKG	51		
	2.2.8	The origin and identity of U-24	53		
2.3	Conclusio	DNS	59		
2.4	Experime	ntal	60		
	2.4.1	General experimental	60		
	2.4.2	UV-visible (UV-vis) absorbance and fluorescence spectrometric measurements	61		
	2.4.3	Reversed phase-high performance liquid chromatography (RP-HPLC)	61		
	2.4.4	Liquid chromatography-mass spectrometry (LC-MS)	62		
	2.4.5	Nuclear magnetic resonance (NMR) spectroscopy			
	2.4.6	Mass spectrometry (MS)			
	2.4.7	Synthesis of 3-hydroxy-2-nitroacetophenone (modified method of Butenandt <i>et al.</i> <sup>263</sup>			
	2.4.8 Synthesis of 4-(3-hydroxy-2-nitrophenyl)-4-oxobut-2-enoic acid (7)				
		2.4.8.1       Method 1 (modified method of Bianchi <i>et al.</i> <sup>276</sup> )			
		2.4.8.2       Method 2 (modified method of Bianchi <i>et al.</i> <sup>276</sup> )			
		2.4.8.3 Method 3			

	2.4.9	Synthesis of methyl 4-(3-hydroxy-2-nitrophenyl)-4-oxobut-2-enoate (10)	. 66
		2.4.9.1 Method 1	66
		2.4.9.1.1 Preparation of dry MeOH <sup>332</sup>	66
		2.4.9.1.2 Trial 1	66
		2.4.9.1.3 Trial 2	67
		2.4.9.1.4 Trial 3	67
		2.4.9.2 Method 2	68
		2.4.9.2.1 Preparation of dry toluene <sup>332</sup>	
		2.4.9.2.2 Trial 1 (modified method of Hermanson <i>et al.</i> <sup>283</sup> )	68
		2.4.9.2.3 Trial 2	68
		2.4.9.3 Method 3	69
		2.4.9.3.1 Preparation of diazomethane <sup>288</sup>	69
		2.4.9.3.2 Trial 1	69
		2.4.9.3.3 Trial 2	69
		2.4.9.3.4 Trial 3	70
		2.4.9.4 Method 4	70
		2.4.9.4.1 Trial 1	70
		2.4.9.4.2 Trial 2	71
	2.4.10	Synthesis of ethyl 4-(3-hydroxy-2-nitrophenyl)-4-oxobut-2-enoate (17)	. 71
		2.4.10.1 Preparation of PPA/SiO <sub>2</sub> catalyst <sup>294</sup>	71
		2.4.10.2 Method 1 (modified method of Sager <i>et al.</i> <sup>293</sup> )	71
		2.4.10.3 Method 2	72
		2.4.10.4 Method 3	73
		2.4.10.5 Method 4	73
	2.4.11	Synthesis of methyl (11) and ethyl (18) 4-(2-nitro-3-((2,3,4,5-tetra-O-acetyl-β-D-	
		glucopyranosyl)oxyphenyl)-4-oxobut-2-enoate	. 73
		2.4.11.1 Preparation of dry DCM <sup>332</sup>	
		2.4.11.2 Method 1	
		2.4.11.3 Method 2	74
		2.4.11.4 Method 3	75
	2.4.12	Synthesis of 3-hydroxykynurenine- <i>O</i> -β-D-glucoside (3OHKG) (modified method of	
		Manthey $et al.^{262}$ )	76
		2.4.12.1 Method 1	
		2.4.12.1 Method 1	
	2.4.13	Stability of 3OHKG and Kyn under basic conditions	
	2.4.14	Stability of 3OHKG under basic conditions in the presence of NH <sub>3</sub>	
	2.4.15	Hydrogenation of U-24	
		2.4.15.1 Method 1	
		2.4.15.2 Method 2	80
СНА	PTER3.		. 81
FACI	LE SYNTF	HESIS OF THE UV FILTER COMPOUNDS 30HKyn AND AHBG	. 81
3.1		ion	
2.1	3.1.1	Previous synthesis of 30HKyn	
	3.1.2	Previous synthesis of AHBG	. 85

3.2	Results	and Discussi	ion					
	3.2.1	Synthesis	of 3-hydroxykynurenine (3OHKyn)					
		3.2.1.1	Synthesis of 2-amino-4-(3-hydroxy-2-nitrophenyl)-4-oxobutanoic acid (40)	-				
			Amination	88				
		3.2.1.2	Synthesis of 3OHKyn - Hydrogenation	89				
		3.2.1.3	Synthesis of 3OHKyn - The "One pot" reaction	89				
	3.2.2	Synthesis	Synthesis of 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid- <i>O</i> -β-D-glucoside (AHBG)					
		90						
		3.2.2.1	Synthesis of 4-(2-amino-3-((2,3,4,5-tetra-O-acetyl-β-D-glucopyranosyl)oxy	phenyl)-				
			4-oxobut-2-anoate (50) - Hydrogenation	90				
		3.2.2.2	Synthesis of AHBG - Deprotection	91				
		3.2.2.3	Synthesis of AHBG – Without intermediate purification	92				
3.3	Conclus	sions		92				
3.4	Experin	nental		93				
	3.4.1	General e	xperimental	93				
	3.4.2	UV-visibl	le (UV-vis) absorbance and fluorescence spectrometric measurements	94				
	3.4.3	Reversed	phase-high performance liquid chromatography (RP-HPLC)	94				
	3.4.4	Liquid ch	romatography-mass spectrometry (LC-MS)	94				
	3.4.5	-	nagnetic resonance (NMR) spectroscopy					
	3.4.6		ctrometry (MS)					
	3.4.7	1	of 3-hydroxykynurenine (3OHKyn)					
	5.1.7	3.4.7.1	Synthesis of 2-amino-4-(3-hydroxy-2-nitrophenyl)-4-oxobutanoic acid (40)					
		5.1.7.1	Amination (modified method of Butenandt <i>et al.</i> <sup>263</sup> )					
		3.4.7.2	Synthesis of 3-hydroxykynurenine (3OHKyn) - Hydrogenation					
		3.4.7.3	Synthesis of 3-hydroxykynurenine (3OHKyn) - The "One pot" reaction					
	3.4.8		of 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid- <i>O</i> -β-D-glucoside (					
	2	97		(1112-0)				
		3.4.8.1	Synthesis of 4-(2-amino-3-((2,3,4,5-tetra-O-acetyl-β-D-glucopyranosyl)oxy	nhenvl)-				
		5.4.0.1	4-oxobutanoate (50) - Hydrogenation					
		3.4.8.1.1	Method 1 (modified method of Manthey <i>et al.</i> <sup>262</sup> )					
		3.4.8.1.2	Method 2 (modified method of Chenault <i>et al.</i> <sup>335</sup> )					
		3.4.8.2	Synthesis of 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid- <i>O</i> -β-D-gluce					
			(AHBG) - Deprotection (modified method of Manthey <i>et al.</i> <sup>262</sup> )					
		3.4.8.3	Synthesis of 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid- <i>O</i> -β-D-gluce					
			(AHBG) - Without intermediate purification (modified method of Manthey					
				99				
CII A	DTED 4			101				
СНА	APIER 4	•••••		101				
SYN	THESIS,	IDENTIFIC/	ATION AND QUANTIFICATION OF NOVEL HUMAN LENS META	BOLITES				
				101				
4.1	Introdu	ction		101				
4.2	Results	and Discussi	ion	102				
	Part A			102				
	Synthes	sis of 4-(2-am	ninophenyl)-4-oxobutanoic acid (AHA)	102				

	4.2.1	Synthesis of 4-(2-nitrophenyl)-4-oxobut-2-enoic acid (56)		
		4.2.1.1 Method 1 and 2		
		4.2.1.2 Method 3		
	4.2.2	Synthesis of 4-(2-aminophenyl)-4-oxobutanoic acid (AHA)	105	
		4.2.2.1 Method 1		
		4.2.2.2 Method 2		
		4.2.2.3 Method 3		
	4.2.3	Synthesis of 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid (AHB)	107	
		4.2.3.1 Method 1		
		4.2.3.2 Method 2		
	4.2.4	Synthesis of glutathionyl-kynurenine (GSH-Kyn) and glutathionyl-3-hydroxyk	ynurenine	
		(GSH-3OHKyn)		
	4.2.5	Synthesis of Kyn yellow and 3OHKyn yellow	111	
		4.2.5.1 Method 1	112	
		4.2.5.2 Alternative methods	113	
		4.2.5.2.1 Method 2	114	
		4.2.5.2.2 Method 3	114	
	4.2.6	Identification of the proposed metabolites in human lenses		
	4.2.7	Stability of the novel lens metabolites	119	
	Part B		125	
	4.2.8	Identification of cysteinyl-3-hydroxykynurenine- <i>O</i> -β-D-glucoside (Cys-3OHK)	G) 125	
	4.2.9	Synthesis and identification of cysteinyl-3-hydroxykynurenine-O-β-D-glucoside (Cys-		
	1.2.9	30HKG)	、 <b>.</b>	
		<ul> <li>4.2.9.1 Synthesis of Cys-3OHKG - Method 1</li> <li>4.2.9.1.1 Synthesis of methyl 2-cysteinyl-4-(2-nitro-3-((2,3,4,5-tetra-<i>O</i>-acetyl-β-D-</li> </ul>	120	
		4.2.9.1.1 Synthesis of mentry 2-cystemy1-4-(2-muo-3-((2,3,4,3-tetra-0-acety1-p-D- glucopyranosyl)oxyphenyl)-4-oxobutanoate (59)	126	
		4.2.9.1.2 Synthesis of methyl 2-cysteinyl-4-(2-amino-3-((2,3,4,5-tetra- <i>O</i> -acetyl-β-D-	120	
		glucopyranosyl)oxyphenyl)-4-oxobutanoate (60)	127	
		4.2.9.1.3 Synthesis of cysteinyl-3-hydroxykynurenine- <i>O</i> -β-D-glucoside (Cys-3OHKC		
		4.2.9.2 Synthesis and stability of cysteinyl-4-oxo-4-phenylbutanoic acid (Cys-B		
		4.2.9.3     Synthesis of Cys-3OHKG - Method 2	<i>,</i>	
		4.2.9.4 Quantification of Cys-3OHKG in human lenses		
		4.2.9.5 Stability of Cys-3OHKG		
4.3	Conclus	ions		
4.4		nental		
	4.4.1	General experimental		
	4.4.2	UV-visible absorbance and fluorescence spectrometric measurements		
	4.4.3	Reversed phase-high performance liquid chromatography (RP-HPLC)		
	4.4.3			
		Liquid chromatography-mass spectrometry (LC-MS)		
	4.4.5	Nuclear magnetic resonance (NMR) spectroscopy		
	4.4.6	Mass spectrometry (MS)		
	4.4.7	Synthesis of 4-(2-nitrophenyl)-4-oxobut-2-enoic acid (56)		
		4.4.7.1 Method 1 (modified method of Bianchi <i>et al.</i> <sup>276</sup> )		
		4.4.7.2 Method 2 (modified method of Bianchi <i>et al.</i> <sup>276</sup> )	141	

		4.4.7.3 Method 3 (modified method of Bianchi <i>et al.</i> <sup>276</sup> )	142
	4.4.8	Synthesis of 4-(2-aminophenyl)-4-oxobutanoic acid (AHA)	142
		4.4.8.1 Method 1	142
		4.4.8.2 Method 2 (modified method of Moriya <i>et al.</i> <sup>338</sup> )	
		4.4.8.3 Method 3 (modified method of Moriya <i>et al.</i> <sup>338</sup> )	
	4.4.9	Synthesis of 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid (AHB)	144
		4.4.9.1 Method 1	
		4.4.9.2 Method 2 (modified method of Moriya <i>et al.</i> <sup>338</sup> )	
	4.4.10	Synthesis of glutathionyl-kynurenine (GSH-Kyn) (modified method of Garner et al. <sup>8</sup>	<sup>6</sup> )145
	4.4.11	Synthesis of glutathionyl-3-hydroxykynurenine (GSH-3OHKyn) (modified method of	
		Garner et al. <sup>86</sup> )	146
	4.4.12	Synthesis of 4-oxo-1,2,3,4-tetrahydroquinoline-2-carboxylic acid (Kyn yellow) (mod	lified
		method of Tokuyama et al. <sup>348</sup> )	147
	4.4.13	Synthesis of 8-hydroxy-4-oxo-1,2,3,4-tetrahydroquinoline-2-carboxylic acid (3OHK	yn
		yellow)	148
		4.4.13.1 Method 1 (modified method of Tokuyama <i>et al.</i> <sup>348</sup> )	148
		4.4.13.2 Method 2 (Bunce <i>et al.</i> <sup>349</sup> )	149
		4.4.13.3 Method 3	150
	4.4.14	Synthesis of cysteinyl-3-hydroxykynurenine-O-β-D-glucoside (Cys-3OHKG)	151
		4.4.14.1 Method 1	151
		4.4.14.1.1 Synthesis of methyl 2-cysteinyl-4-(2-nitro-3-((2,3,4,5-tetra-O-acetyl-β-D-	
		glucopyranosyl)oxyphenyl)-4-oxobutanoate (59)	151
		4.4.14.1.2 Synthesis of methyl 2-cysteinyl-4-(2-amino-3-((2,3,4,5-tetra-O-acetyl-β-D-	
		glucopyranosyl)oxyphenyl)-4-oxobutanoate (60)	151
		4.4.14.1.3 Synthesis of cysteinyl-3-hydroxykynurenine- <i>O</i> -β-D-glucoside (Cys-3OHKG)	
		4.4.14.2 Method 2 (modified method of Garner <i>et al.</i> <sup>86</sup> )	
		4.4.14.2.1 Synthesis of cysteinyl-3-hydroxykynurenine- <i>O</i> -β-D-glucoside (Cys-3OHKG)	153
	4.4.15	Synthesis of cysteinyl-β-benzoylbutanoic acid (Cys-BAA)	154
	4.4.16	Lens preparation and RP-HPLC purification	154
	4.4.17	Quantification of lens UV filters and their metabolites by LC-MS and RP-HPLC	155
	4.4.18	Extraction efficiency of Kyn, 3OHKyn, 3OHKG, AHA, AHB, GSH-Kyn and GSH-	
		3OHKyn from lens proteins	156
	4.4.19	Stability of AHA, AHB, GSH-Kyn and GSH-3OHKyn under extraction, HPLC and	рН 7.0
		(aerobic) conditions	156
	4.4.20	Stability of AHB, GSH-Kyn and GSH-3OHKyn at pH 7.0 (anaerobic)	157
	4.4.21	Stability of Cys-BAA	
	4.4.22	Stability of Cys-3OHKG at pH 7.0	
HAI	PTER 5.		159
IOD	IFICATIO	N OF BOVINE LENS PROTEIN WITH UV FILTERS AND RELATED METABOL	ITES
			159
.1	Introduct	ion	
.2		nd Discussion	
. —	5.2.1	Modification of BLP with Kyn, 30HKyn, 30HKG, AHB and AHA	

	5.2.2	Characterisation of UV filter-treated BLP by UV-vis absorbance and fluorescence.	162
	5.2.3	RP-HPLC analysis of modified BLP	163
	5.2.4	Synthesis of 3OHKyn amino acid adducts	164
		5.2.4.1 Synthesis of <i>t</i> -Boc-His-, <i>t</i> -Boc-Lys- and Cys-3OHKyn	164
		5.2.4.2 Synthesis of His- and Lys-3OHKyn	167
	5.2.5	Acid hydrolysis of UV filter-treated and unmodified BLP	168
		5.2.5.1 Untreated BLP	
		5.2.5.2 Kyn-treated BLP	
		5.2.5.3 30HKyn-treated BLP	
		5.2.5.4 30HKG-treated BLP	
		<ul><li>5.2.5.5 AHB-treated BLP</li><li>5.2.5.6 AHA-treated BLP</li></ul>	
	5.2.6	Removal of non-covalently bound material from modified proteins	
	5.2.7	Stability of UV filters and their adducts during acid hydrolysis	
	5.2.7	Stability of AHA and AHB at pH 9.5	
5.2	5.2.9	Characterisation of low molecular mass compounds isolated upon dialysis	
5.3		ions	
5.4	1	iental	
	5.4.1	General experimental	
	5.4.2	UV-visible (UV-vis) absorbance and fluorescence spectrometric measurements	
	5.4.3	Reversed phase-high performance liquid chromatography (RP-HPLC)	
	5.4.4	Liquid chromatography-mass spectrometry (LC-MS)	
	5.4.5	Nuclear magnetic resonance (NMR) spectroscopy	
	5.4.6	Mass spectrometry (MS)	
	5.4.7	Synthesis of 3OHKyn amino acid adduct (modified method of Vazquez et al. <sup>18</sup> )	
	5.4.8	Acid hydrolysis of $N$ - $\alpha$ - $t$ -Boc protected 3OHKyn amino acid adducts (modified mo	
		Vazquez <i>et al.</i> <sup>18</sup> )	
	5.4.9	Preparation of BLP <sup>117</sup>	
	5.4.10	Modification of BLP with UV filters <sup>18,247</sup>	
	5.4.11	Acid hydrolysis of UV filter-treated proteins <sup>20</sup>	197
	5.4.12	Stability of AHA at pH 9.5	197
	5.4.13	Stability of AHB at pH 9.5	198
	5.4.14	Extraction of unbound UV filters from modified proteins	198
	5.4.15	Stability of UV filters and amino acid adducts under acid hydrolysis	198
СНА	APTER 6		199
EFFI	ECT OF U	V LIGHT ON UV FILTER-TREATED LENS PROTEINS	199
6.1	Introduc	tion	199
6.2	Results	and Discussion	200
	Part A		201
	6.2.1	Determination of peroxides	201
		6.2.1.1 Kyn-, 3OHKyn- and 3OHKG-modified bovine lens proteins	202

		6.2.1.2	Determination of peroxide yield-dependency on the level of protein-bo	und 3OHKG
			and 3OHKyn	
		6.2.1.3	Determination of wavelength-dependency	
		6.2.1.4	Determination of O <sub>2</sub> -dependency	
		6.2.1.5	Determination of <sup>1</sup> O <sub>2</sub> role	
		6.2.1.6	Role of free 3OHKyn and 3OHKG and their Cys adducts in peroxide f	ormation.209
		6.2.1.7	Role of AHB- and AHA-treated BLP and free AHB and AHA in perox	tide
			formation	
		6.2.1.8	LC-MS investigation of illuminated UV filters	
	Part B			
	6.2.2	Oxidation	products of Tyr (DOPA and di-Tyr) and Phe (o- and m-Tyr)	
	6.2.3	Determina	ation of DOPA and di-Tyr	
		6.2.3.1	Kyn-modified BLP	
		6.2.3.2	3OHKG-modified BLP	
		6.2.3.3	3OHKyn-modified BLP	221
		6.2.3.4	Determination of O <sub>2</sub> -dependency	
		6.2.3.5	Determination of <sup>1</sup> O <sub>2</sub> role	
		6.2.3.6	AHB- and AHA-treated BLP	
	Part C			
	6.2.4	Determina	ation of protein cross-linking	
6.3	Conclus	ions		
6.4	Experim	ental		
	6.4.1	General e	xperimental	
	6.4.2	Illuminati	on procedures	
	6.4.3	Peroxide of	quantification	
	6.4.4	Liquid ch	romatography-mass spectrometry (LC-MS)	
	6.4.5	-	id analysis	
	6.4.6		odecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE)	
	6.4.7		analysis	
			-	
СНА	PTER 7	•••••		
CON	CLUSION	IS AND FU	TURE DIRECTIONS	
REF	ERENCE	S		
APPI	ENDIX A	•••••		279
APPI	ENDIX B	•••••		
APPI	ENDIX C	••••••		
MED		DAGE		200
NED	IA CUVE	KAUĽ		

# LIST OF FIGURES

Figure 1.1: The cross-section of the human eye. <sup>22</sup>	3
Figure 1.2: The cross-section of the human lens (adapted with modifications from Wormstone <i>et</i> and Harding <sup>24</sup> ).	<i>al.</i> <sup>25</sup> 4
Figure 1.3: Chemical structure of GSH.	6
Figure 1.4: Chemical structure of ascorbic acid.	7
Figure 1.5: The major UV filters in the human lens.	9
Figure 1.6: Selected AGE products found in human lenses.	16
Figure 1.7: The Japanese Bridge painted by Claude Monet in 1899 (top) and 1922 (bottom). <sup>180</sup>	18
Figure 1.8: Pirie's classification of nuclear cataract using slit lamp photographs (adapted Duncan <sup>64</sup> ).	from 19
Figure 1.9: Possible oxidation products of 3OHKyn: xanthommatin (1), dihydroxanthommatin (2), dihydroxyquinolinequinone carboxylic acid (3), and benzoxazole (4) and benzimidazol adducts. R = amino acid or peptide.	
Figure 1.10: Diagram of the eye illustrating the wavelengths of light that are transmitted and absorby the human eye (adapted with modifications from Sliney <sup>226</sup> ).	orbed 22
Figure 2.1: The products seen upon nitration of 3-hydroxyacetophenone by "Menke conditions hydroxy-2-nitroacetophenone (9); 3-hydroxy-4-nitroacetophenone (12); 3-hydrox nitroacetophenone (13).	
Figure 2.2: Temperature (blue), pressure (green) and power (red) measurements during the synthes the acrylic acid (7) in the cavity of a focused microwave reactor (CEM Discover) at 110°C	
Figure 2.3: Stability of 3OHKG in aqueous NaOH (0.05 M) at pH 12.5 in the absence of NH <sub>3</sub> (♠) in the presence of NH <sub>3</sub> (0.04 M) (●).	) and 53
Figure 2.4: Proposed structure of U-24.	55
Figure 2.5: ES-MS/MS of authentic 3OHKG (A) and U-24 (B) in positive mode.	56
Figure 2.6: 3OHKG (A) and the proposed structure of U-24 (B) with the assigned major proton fragments by ES-MS/MS (positive mode).	nated 57
	51

- Figure 4.1: RP-HPLC profile of the UV filter extract from a normal lens nucleus (88 years old). Detection was at 360 nm in Absorbance units (arbitrary units). 3OHKG (1); Kyn (2); GSH-3OHKG (3); AHBG (4) and AHBDG (5). "X" represents an unknown compound shown as a double peak on the RP-HPLC trace. Expected elution times of AHB and GSH-3OHKyn (6), AHA and GSH-Kyn (7), 3OHKyn yellow (8) and Kyn yellow (9).
- Figure 4.2: The colour of solutions of AHB in PBS (pH 7.0) at 37°C at time 0 h (A) and after 200 h (B) under aerobic conditions. 121
- Figure 4.3: AHB (0.25 mM) was incubated in PBS (pH 7.0) in the presence of atmospheric oxygen (♦), in oxygen free PBS (pH 7.0) in the presence of BHT (0.1 mM) (Δ) and in oxygen free PBS (pH 7.0) in the presence of ascorbic acid (2.0 mM) and BHT (0.1 mM) (■). Aliquots of the reaction mixtures were taken at the indicated time points and analysed by RP-HPLC. Detection was at 254 nm.
- Figure 4.4: GSH-Kyn (0.40 mM) was incubated in PBS (pH 7.0) in the presence of atmospheric oxygen or in oxygen free PBS (pH 7.0) in the presence of ascorbic acid (0.2 mM) and BHT (0.1 mM). Aliquots of the reaction mixtures were taken at the indicated time points and analysed by RP-HPLC. Detection was at 254 nm. GSH-Kyn (♦); deaminated Kyn (■); Kyn yellow (▲).
- Figure 4.5: GSH-3OHKyn (0.10 mM) was incubated in PBS (pH 7.0) in the presence of atmospheric oxygen or in oxygen free PBS (pH 7.0) in the presence of ascorbic acid (2.0 mM) and BHT (0.1 mM). Aliquots of the reaction mixtures were taken at the indicated time points and analysed by RP-HPLC. Detection was at 254 nm. GSH-3OHKyn (♦); deaminated 3OHKyn (■); 3OHKyn yellow (▲).
- Figure 4.6: Stability of Cys-BAA at pH 2.5 (♦), 4.0 (■), 7.2 (▲), 9.5 (◊), 12.5 (□) and 12.5 (excess of Cys, 0.15 mol equivalents; ●) for 25 h. 130
- Figure 4.7: Proposed dimeric structure (*m*/*z* 474 (M+H<sup>+</sup>)) formed following incubation of Cys-BAA at pH 9.5 for 25 h. 130
- Figure 4.8: ES-MS/MS of compound (X) isolated from human lenses (A) and synthetic Cys-3OHKG (B) in positive ion mode. Mass differences between major ions are indicated in Daltons (Da).132
- Figure 4.9: Quantification of Cys-3OHKG in normal human lenses. Cys-3OHKG was quantified in the nucleus (shaded) and cortex (unshaded) of normal human lenses. 133
- Figure 4.10: Linear regression plot of the nuclear ( $\blacklozenge$ ,  $R^2 = 0.0439$ ) and cortical ( $\circ$ ,  $R^2 = 0.1823$ ) concentration of Cys-3OHKG versus 3OHKG in normal human lenses. The quantity of each compound was determined by comparison to a standard curve. 134

- Figure 4.11: Linear regression plot of the nuclear ( $\blacklozenge$ ,  $R^2 = 0.9105$ ) and cortical ( $\circ$ ,  $R^2 = 0.7033$ ) concentration of Cys-3OHKG versus GSH-3OHKG in normal human lenses. The quantity of each compound was determined by comparison to a standard curve. 135
- Figure 4.12: Cys-3OHKG (0.2 mM) was incubated in argon-gassed PBS (pH 7.0). Aliquots of the reaction mixtures were taken at the indicated time points and analysed by RP-HPLC. Detection was at 360 nm. Cys-3OHKG (♦); deaminated 3OHKG (■).
  136
- Figure 4.13: Chemical structures of the decysteinated 3OHKG (23) and 3OHKG yellow. 137
- Figure 5.1: The reaction mixtures of BLP treated with the UV filters before dialysis. 3OHKG at pH 9.5 and 7.2, 3OHKyn at pH 9.5 and 7.2, AHA at pH 9.5 and 7.2, and AHB at pH 9.5 and 7.2, from left to right.
- Figure 5.2: Lyophilised UV filter-treated BLP after dialysis. 3OHKG at pH 9.5 and 7.2, 3OHKyn at pH 9.5 and 7.2, AHA at pH 9.5 and 7.2, and AHB at pH 9.5 and 7.2, from left to right.
- Figure 5.3: Characteristic fragments observed by ES-MS/MS analysis (positive mode) of 3OHKyn amino acid adducts. 166
- Figure 5.4: Chemical structures of His- and Lys-3OHKyn. 167
- Figure 5.5: RP-HPLC profiles of acid-hydrolysed unmodified BLP. BLP incubated at pH 7.2 for 14 days (A); BLP incubated at pH 9.5 for 48 h (B).
- Figure 5.6: RP-HPLC profile of acid-hydrolysed Kyn-treated BLP at pH 9.5 for 48 h at 37°C. Lys-Kyn (1); diastereomers of Cys-Kyn (2, double peak); unidentified peak (3). 171
- Figure 5.7: RP-HPLC profiles of acid-hydrolysed 3OHKyn-treated BLP at pH 7.2 (A) and pH 9.5 (B) for 48 h at 37°C. Lys-3OHKyn (1); diastereomers of Cys-3OHKyn (2); 35.0 min (3) and 36.0 min (4) peaks were unidentified.
- Figure 5.8: RP-HPLC profiles of acid-hydrolysed 3OHKG-treated BLP at pH 7.2 (A) for 14 days and pH 9.5 (B) for 48 h at 37°C. Lys-3OHKyn (1); diastereomers of Cys-3OHKyn (2); 35.0 min (3) and 36.0 min (4) peaks were unidentified.
- Figure 5.9: RP-HPLC profiles of acid-hydrolysed AHB-treated BLP at pH 7.2 (A) and pH 9.5 (B) for 48 h at 37°C. 31.4 min (1, AHB); 37.6 min (2); 38.6 min (3); 39.5 min (4). 177
- Figure 5.10: RP-HPLC profiles of acid-hydrolysed AHA-treated BLP at pH 7.2 (A) and pH 9.5 (B) for 48 h at 37°C. 178
- Figure 5.11: RP-HPLC profiles at 360 nm and 254 nm (inset) of acid-hydrolysed 30HKG (0.6 mg) (A) and 30HKG (0.4 mg) in the presence of BLP (13.2 mg) (B). 30HKyn (1); decomposition

product of m/z 461 (M+H<sup>+</sup>) and  $\lambda_{max}$  263/367 nm (2, double peak). RP-HPLC injection volumes were 50  $\mu$ L for A and B. 182

- Figure 5.12: RP-HPLC profiles at 360 nm and 254 nm (inset) of acid-hydrolysed Cys-3OHKG (1.1 mg) (A) and Cys-3OHKG (0.5 mg) in the presence of BLP (10.5 mg) (B). Cys-3OHKyn (1, double peak); unknown decomposition product of m/z 300 (M+H<sup>+</sup>) and  $\lambda_{max}$  266/372 nm (2); unknown decomposition product of m/z 264 (M+H<sup>+</sup>) and  $\lambda_{max}$  252/340 nm (3). RP-HPLC injection volumes were 5  $\mu$ L (A) and 50  $\mu$ L (B).
- Figure 5.13: RP-HPLC profiles at 360 nm and 254 nm (inset) of acid-hydrolysed AHB (0.3 mg) (A) and AHB (0.2 mg) in the presence of BLP (13.1 mg) (B). AHB (1). RP-HPLC injection volumes were 5 μL (A) and 50 μL (B).
- Figure 5.14: Stability of AHB over 48 h at pH 9.5 at 37°C. (A) Duplicate analysis and (B) single analysis of AHB stability with acidification of samples before storage or RP-HPLC analysis. AHB (■); 30.7 min (m/z 413 (M+H<sup>+</sup>)) (▲); 31.7 min (m/z 399 (M+H<sup>+</sup>)) (●); 32.7 (m/z 414 (M+H<sup>+</sup>)) (●).
- Figure 6.1: Concentration of peroxides generated following illumination of Kyn-modified lens protein (0.96 mol of Cys-Kyn per mol protein, ) at 4°C and pH ~7 and corresponding non-illuminated control (); 3OHKG-modified lens protein (0.74 mol of Cys-3OHKyn per mol protein, ) and corresponding non-illuminated control (); 3OHKyn-modified lens protein (0.05 mol of Cys-3OHKyn per mol protein, ) and corresponding non-illuminated control (); and unmodified lens proteins () and corresponding non-illuminated control (); and unmodified lens protein (1 mg/mL) were used in each case. Data are means ± SD of triplicate measurements from a single experiment typical of several.
- Figure 6.2: Effect of the level of UV filter modification on peroxide yield. Concentration of peroxides generated following illumination of A) 3OHKG-modified lens proteins (0.40 mol of Cys-3OHKyn per mol protein; white bar, 0.78 mol of Cys-3OHKyn per mol protein; striped bar, and 1.14 mol of Cys-3OHKyn per mol protein; dotted bar) and B) 3OHKyn-modified lens proteins (0.05 mol of Cys-3OHKyn per mol protein; white bar, 0.08 mol of Cys-3OHKyn per mol protein; striped bar, and 0.64 mol of Cys-3OHKyn per mol protein; dotted bar) at 4°C and pH ~7. Equal concentrations of protein (1 mg/mL) were used in each case. Data are means ± SD of triplicate measurements from a single experiment typical of several.
- Figure 6.3: Effect of wavelength of illuminating light on peroxide production by 3OHKG-modified lens proteins (0.74 mol of Cys-3OHKyn per mol protein, 1 mg protein/mL) at 4°C and pH ~7. Illumination was carried out as described in the Experimental section using filters which cut-off the transmitted light at 305 (♦), 345 (■) and 385 (▲) nm. Data are means ± SD of triplicate measurements from a single experiment typical of several.

- Figure 6.4: Effect of air (◆) and N<sub>2</sub> (■) atmospheres on the yield of peroxides formed at 4°C and pH ~7 during illumination of A) 3OHKG-modified lens proteins (1 mg protein/mL, 0.74 mol of Cys-3OHKyn per mol protein) and B) 3OHKyn-modified lens protein (1 mg protein/mL, 0.05 mol of Cys-3OHKyn per mol protein). Modified protein samples illuminated under an air atmosphere were statistically different to the samples illuminated under N<sub>2</sub> at the 60 and 120 min time points as assessed by Student's *t*-test. Data are means ± SD of triplicate measurements from a single experiment typical of several.
- Figure 6.5: Effect of buffers made up using H<sub>2</sub>O (white and striped bars) versus D<sub>2</sub>O (black and dotted bars) on peroxide formation at 4°C and pH / pD 7, from A) 3OHKG-modified lens protein (0.40 mol of Cys-3OHKyn per mol protein) and B) 3OHKyn-modified lens protein (0.05 mol of Cys-3OHKyn per mol protein). Catalase was added to some of the samples immediately after the cessation of illumination (striped and dotted bars); controls (white and black bars) did not have catalase added. 3OHKG-modified lens proteins (2 mg/mL) and 3OHKyn-modified lens proteins (0.5 mg/mL) made up in D<sub>2</sub>O were statistically different from the samples made up in H<sub>2</sub>O at the 120 min time point as assessed by one-way ANOVA with Tukey's post hoc-test. The samples made up in D<sub>2</sub>O and H<sub>2</sub>O at the 0 min time point were not significantly different from the illuminated samples with added catalase at the 120 min time point. Data are means ± SD of triplicate measurements from a single experiment typical of several.
- Figure 6.6: Effect of sodium azide (10 mM; ◆,■) on the formation of peroxides following illumination of 3OHKyn-modified lens protein (1 mg protein/mL, 0.05 mol of Cys-3OHKyn per mol protein) compared to samples with no added sodium azide (▲,●) at 4°C and pH ~7. Control samples, with (■) or without (●) azide were incubated in the dark for 120 min. Data in the presence of sodium azide were statistically different to the samples in the absence of sodium azide. Both illuminated samples were statistically different from the non-illuminated samples in either the presence or absence of sodium azide, as assessed by one-way ANOVA with Tukey's post hoc-test. Data are means ± SD of triplicate measurements from a single experiment typical of several.
- Figure 6.7: Effect of protein-bound versus protein-free UV filters on peroxide yield. Peroxide formation following illumination of A) 3OHKG-modified lens proteins (1 mg protein/mL, 0.78 mol of Cys-3OHKyn per mol protein; white bars) and B) 3OHKyn-modified lens proteins (1 mg protein/mL, 0.64 mol of Cys-3OHKyn per mol protein; white bars) compared to unmodified lens proteins with added free UV filters and free Cys amino acid adducts. Unmodified lens proteins (1 mg protein/mL, dotted bars) were illuminated in the presence of A) free 3OHKG (38.9 μM, striped bars) or free Cys-3OHKG adduct (38.9 μM, black bars), and B) free 3OHKyn (31.7 μM, striped bars) or free Cys-3OHKyn adduct (31.7 μM, black bars). Illuminated 3OHKG- and 3OHKyn-modified lens proteins samples were statistically different from the unmodified lens proteins in the presence or absence of free UV filters or

free Cys-UV filter adducts as assessed by one-way ANOVA with Tukey's post hoc-test. Data are means  $\pm$  SD of triplicate measurements from a single experiment typical of several. 210

- Figure 6.8: Peroxide formation from free 3OHKyn (white and striped bars) and free 3OHKG (black and dotted bars) (4.5 mM and 2.6 mM, respectively, in H<sub>2</sub>O at 4°C and pH ~7), following illumination (white and black bars) or incubation in the dark (striped and dotted bars) for 120 min. The peroxide levels detected with illuminated 3OHKyn were statistically different from the non-illuminated control at 60 and 120 min time points as assessed by Student's *t*-test. The peroxide levels detected from the illuminated and control 3OHKG samples were not statistically different. Data are means  $\pm$  SD of triplicate measurements from a single experiment typical of several. 211
- Figure 6.9: A) Peroxide formation following illumination of AHB-modified lens proteins (0.09 mol of AHB per mol protein; ■) and non-illuminated controls (■) and illuminated unmodified lens proteins, with added free AHB (4.7 µM; ■) and corresponding non-illuminated control (■), compared to illuminated unmodified lens proteins (■) and non-illuminated unmodified lens protein (■). The protein concentration in each case was 1 mg protein/mL. The peroxide levels detected from the illuminated AHB-modified lens protein samples were not statistically different from the corresponding non-illuminated controls as assessed by Student's *t*-test. B) Peroxide formation from free AHB (4.8 mM) over 120 min of illuminated AHB samples were statistically different from the non-illuminated controls at the 120 min time point as assessed by Student's *t*-test. Data are means ± SD of triplicate measurements from a single experiment typical of several.
- Figure 6.10: Typical RP-HPLC trace of acid hydrolysates of 3OHKG-modified proteins (0.78 mol of Cys-3OHKyn per mol protein) after 120 min of illumination. The level of parent Tyr (Rt 14.3 min) was quantified using UV detection (λ 280 nm) (A), and Tyr oxidation products were quantified by fluorescence detection (λ<sub>ex/em</sub> 280/320 nm for DOPA (Rt 9.5 min) and λ<sub>ex/em</sub> 280/410 nm for di-Tyr (Rt 34.3 min)) (B). *m*-Tyr (Rt 16.1 min) and *o*-Tyr (Rt 21.2 min) visibly coeluting with unknown peaks.
- Figure 6.11: Time course of formation of DOPA (A) and di-Tyr (B) during illumination of Kynmodified lens proteins (0.96 mol of Cys-Kyn per mol protein, white bars) and unmodified proteins (black bars). Samples of Kyn-modified (striped bars) and unmodified lens proteins (dotted bars) were kept in the dark as controls. Data (expressed as mM of modified amino acid per mol of parent Tyr) are means ± SD of triplicate (0, 15, 30 min time points) and quadruplicate (60 and 120 min time points) samples. For both measurements, DOPA (A) and di-Tyr (B) quantities were statistically elevated from those for the control samples kept in the dark, and unmodified proteins as assessed by one-way ANOVA with Tukey's post hoc-test. 220

- Figure 6.12: Time course of formation of di-Tyr during illumination of 3OHKG-modified lens proteins (0.78 mol of Cys-3OHKyn per mol protein, ◆) and unmodified lens proteins (■). Simultaneously, control samples that were kept in the dark, of modified and unmodified proteins were also analysed. Data (expressed as mM of modified amino acid per mol of parent Tyr) are means ± SD of triplicate (15 and 30 min time points) or quadruplicate (0, 60 and 120 min time points) experiments with the values from control samples subtracted (samples kept in the dark). Di-Tyr concentrations were statistically elevated from those of the control samples as assessed by Student's t-test.
- Figure 6.13: Time course of formation of DOPA during illumination of 3OHKyn-modified lens proteins (0.64 mol of Cys-3OHKyn per mol protein, white bars) and unmodified lens proteins (black bars). Control samples were kept in the dark, modified (striped bars) and unmodified (dotted bars) proteins were also analysed. Data (expressed as mM of modified amino acid per mol of parent Tyr) are means ± SD of triplicate (0, 15, 30 and 60 min time points) and quadruplicate (120 min time point) samples. DOPA quantities were statistically-elevated from those of the unmodified samples as assessed by one-way ANOVA with Tukey's post hoc-test. The levels of DOPA detected on illuminated 3OHKyn-modified samples were not statistically different from the non-illuminated control samples at any time point.
- Figure 6.14: Time course of formation of protein-bound DOPA (A) and di-Tyr (B) on illumination of 3OHKG-modified lens proteins (1 mg protein/mL, 0.78 mol of Cys-3OHKyn per mol protein, white bars) and 3OHKyn-modified lens protein (1 mg protein/mL, 0.64 mol of Cys-3OHKyn per mol protein, black bars) under an atmosphere of air versus N<sub>2</sub>. Control samples of 3OHKG-modified lens protein (striped bars) and 3OHKyn-modified lens protein (dotted bars) kept under N<sub>2</sub> were also analysed. Data (expressed as mM of modified amino acid per mol of parent Tyr) are means ± SD of triplicate (15 and 30 min time points) and quadruplicate (0, 60 and 120 min time points) samples. In (A), the levels of the oxidised products detected under air versus N<sub>2</sub> were not statistically different for all of the samples at any time point. In (B), the levels detected in the air versus N<sub>2</sub>-gassed samples were statistically different at the 30, 60 and 120 min time points for the 3OHKG-modified lens protein samples, but not for 3OHKyn-modified lens protein samples as assessed by Student's *t*-test.
- Figure 6.15: Time course of Tyr levels from 3OHKG (0.40 mol of Cys-3OHKyn per mol protein; A) and 3OHKyn (0.05 mol of Cys-3OHKyn per mol protein; B) modified lens protein samples upon illumination. Samples were prepared in H<sub>2</sub>O (illuminated control, ◆) in the presence of sodium azide (■), or prepared in D<sub>2</sub>O (▲). A control sample of unmodified lens protein prepared in H<sub>2</sub>O was analysed simultaneously (●). Data (expressed as pmol of parent Tyr) are means ± SD of triplicate (0, 15 and 30 min time points) or quadruplicate (60 and 120 min time points) samples.
- Figure 6.16: Time course of DOPA levels from AHB (0.09 mol of AHB per mol protein, white and striped bars) and AHA (0.11 mol of AHA per mol protein, black and dotted bars) modified lens protein samples upon illumination. Control samples for AHB (striped bars) and AHA

(dotted bars) were kept in the dark and analysed simultaneously. Data (expressed as mM of modified amino acid per mol of parent Tyr) are means  $\pm$  SD of triplicate (0, 15 and 30 min time points) and quadruplicate (60 and 120 min time points) samples. In both observations (AHB- and AHA-treated lens proteins) there was an UV light independent steady increase in DOPA levels. 225

- Figure 6.17: SDS-PAGE of lens proteins treated with Kyn (0.96 mol Cys-Kyn per mol protein), 3OHKyn (0.64 mol Cys-3OHKyn per mol protein), 3OHKG (0.78 mol Cys-3OHKyn per mol protein), AHA (0.11 mol AHA per mol protein) and AHB (0.09 mol AHB per mol protein) at pH 9.5 for 48 h, together with molecular mass markers, analysed on a 12% Tris-Glycine gels. Protein bands were detected by silver staining.
- Figure 6.18: Spectral irradiance of light from 125 W mercury arc lamp. A) no filter; B) 305 nm filter; C) 345 nm filter; D) 385 nm filter. 230
- Figure 6.19: Solutions of BLP were illuminated using a system as shown in the diagram. The samples were prepared in 20 mL glass vials (3) and placed at 4°C (2) behind the 305, 345 or 385 nm cut-off filter (4). Samples were at a distance of 10 cm (5) from a broad spectrum 125 W mercury arc lamp (6). Samples were gassed continuously during the illumination using compressed air or nitrogen gas (1).

### LIST OF SCHEMES

Scheme 1.1: The glutathione protective system. <sup>64,68</sup>	7
Scheme 1.2: The role of ascorbic acid as antioxidant and pro-oxidant in the antioxidative pathway the human lens. <sup>64,72,77,78</sup>	ys in 8
Scheme 1.3: The kynurenine pathway of tryptophan metabolism. <sup>94-96,99</sup>	10
Scheme 1.4: The biosynthetic pathway for formation of AHBG and GSH-3OHKG. <sup>85,86,100</sup>	10
Scheme 1.5: Binding of 3OHKG, Kyn and 3OHKyn to nucleophilic amino acid residues.	17
Scheme 2.1: Synthesis of 3OHKG from 3OHKyn by <sup>a</sup> Heckathorn <i>et al.</i> <sup>260</sup> and <sup>b</sup> Real and Ferré. <sup>261</sup>	27
Scheme 2.2: Synthesis of 3OHKG by Manthey et al. <sup>262</sup>	28
Scheme 2.3: The mechanism of electrophilic nitration of an aromatic ring by "Menke conditions". <sup>24</sup>	66,267 29
Scheme 2.4: Synthetic pathway towards the aldol (14), the acrylic acid (7) and the ester (15).	33
Scheme 2.5: Formation of the methoxy methyl ester (16) via Michael addition of MeOH.	37
Scheme 2.6: Mechanism for acid catalysed demethanolysis of the methoxy methyl ester (16) to methyl ester (10).	the 38
Scheme 2.7: The mechanism for methylation of the carboxylic acid of the acrylic acid (7) u diazomethane.	sing 38
Scheme 2.8: Koenigs-Knorr reaction mechanism. A, homogenous process (catalysed by sol catalysts); B, heterogeneous process (catalysed by insoluble catalysts). <sup>310-313</sup>	uble 45
Scheme 2.9: Proposed mechanism for deamination of 3OHKG at pH ~7-9.5.	52
Scheme 2.10: Reduction of U-24 to 3OHKG.	58
Scheme 2.11: The reduction of ethyl 1-acetyl-3-(5-methoxy-2-nitrobenzoyl)- $\Delta^2$ -pyrazolin carboxylate (24) to ethyl 1-acetyl-3-(5-methoxybenzo[c]isoxazol-3-yl)- $\Delta^2$ -pyrazolin carboxylate (25). <sup>325</sup>	
Scheme 3.1: Synthesis of 3OHKyn by Kotake <i>et al.</i> <sup>333</sup>	82
Scheme 3.2: Synthesis of 3OHKyn by Butenandt <i>et al.</i> <sup>263</sup>	83
Scheme 3.3: Synthesis of 3OHKyn by Brown <i>et al.</i> <sup>334</sup>	84

Scheme 3.4: Synthesis of AHBG by Chenault <i>et al.</i> <sup>335</sup>	85
Scheme 3.5: Synthesis of AHBG by Manthey <i>et al.</i> <sup>262</sup>	86
Scheme 3.6: Proposed synthetic pathway for 3OHKyn and AHBG from the acrylic acid (7) and methyl ester glucoside (11), respectively.	d the 87
Scheme 4.1: Proposed <i>in vivo</i> degradation pathways of the human lens UV filters Kyn, 3OHKyn 3OHKG.	and 102
Scheme 4.2: Synthesis of AHA by Rossi <i>et al.</i> <sup>337</sup>	103
Scheme 4.3: Proposed synthetic pathway towards AHA.	103
Scheme 4.4: Proposed pathways leading to i) intra- and ii) intermolecular Michael adducts in synthesis of AHA.	the 105
Scheme 4.5: Proposed pathways leading to intramolecular Michael adducts in the synthesis of A	HB.
The tautomeric forms of xanthurenic acid.	109
Scheme 4.6: Formation of GSH-Kyn and GSH-3OHKyn under basic conditions.	111
Scheme 4.7: Formation of Kyn yellow and 3OHKyn yellow under basic conditions.	112
Scheme 4.8: Oxidation of Kyn yellow and 3OHKyn yellow to kynurenic acid and xanthurenic a	acid,
respectively.	113
Scheme 4.9: Ring closure by tandem reduction-Michael addition by Bunce et al. <sup>349</sup>	114
Scheme 4.10: Proposed pathway for synthesis of 3OHKyn yellow from the acrylic acid (7).	115
Scheme 4.11: Proposed mechanism for the oxidation of AHB at pH 7.0 (PBS) under aerobic condit	tions
and formation of the oxidation product of molecular mass of 412 Da. <sup>201,209,362</sup>	122
Scheme 4.12: Proposed decomposition products of AHB at pH 7.0 (PBS) under aerobic conditions.	122
Scheme 4.13: Proposed synthetic pathway towards Cys-3OHKG.	126
Scheme 5.1: Deamination of UV filters and binding to lens proteins.	159
Scheme 5.2: Acid hydrolysis of Kyn-, 3OHKyn- and 3OHKG-treated BLP, and recovery of Kyn 3OHKyn-amino acid adducts.	-and 164
Scheme 5.3: Proposed mechanism for the formation of the <i>t</i> -Boc-His adduct of 3OHKyn.	165
Scheme 5.4: Proposed formation of the quinone dimer (62). <sup>392</sup>	183

- Scheme 6.2: Oxidation of Tyr by HO in the absence of O<sub>2</sub> (Path A) and presence of O<sub>2</sub> (Path B) to yield DOPA.<sup>359</sup> 216
- Scheme 6.3: Tyrosine oxidation involves one-electron oxidation to form phenoxyl (tyrosyl) radicals (Tyr) that can dimerise to give both carbon-oxygen (Path A) and carbon-carbon (Path B) linked dimers.<sup>359,420</sup> 216
- Scheme 6.4: Oxidation of Phe by HO in the presence of  $O_2$  (Path A) and absence of  $O_2$  (Path B). The position of the hydroxyl group on Phe can be at *o*-, *m* and *p* position.<sup>359</sup> 217
- Scheme 7.1: The improved synthetic strategy towards 3OHKG. The preferred reaction conditions and product yields: i) Cu(NO<sub>3</sub>)<sub>2</sub>·2.5H<sub>2</sub>O, AcO<sub>2</sub>, AcOH, 10-15°C, 16 h, 25%; ii) HCOCOOH, MW, 110°C, 17 min, 60%; iii) MeOH, H<sub>2</sub>SO<sub>4</sub>, reflux, 6 h, 90%; iv) HCOCOOEt (50% sol. in toluene), PPA/SiO<sub>2</sub>, MW, 110°C, 130 min, 58%; v) ABG, DCM, K<sub>2</sub>CO<sub>3</sub>, tetra-*n*-butyl ammonium bromide, 2.5 days, 70-78%; vi) EtOAc, aqueous NH<sub>3</sub>, 3 h, not isolated; vii) EtOAc, H<sub>2</sub>, Pd/C, 3 h, not isolated; viii) aqueous NaOH, NH<sub>3</sub>, 8.5 h, 27% 3OHKG and 18-20% U-24; ix) H<sub>2</sub>O, H<sub>2</sub>, Pd/C, 10 min, 92%.
- Scheme 7.2: The improved synthetic strategy towards 3OHKyn (A) and AHBG (B). The preferred reaction conditions and product yields: A, i) aqueous NH<sub>3</sub>, 1.5 h, not isolated; ii) aqueous NH<sub>3</sub>, H<sub>2</sub>, Pd/C, 2 h, 72%; B, i) H<sub>2</sub>, Pd/C, EtOAc/EtOH (4:1, v/v), 2 h, not isolated; ii) aqueous NaOH, 5 h, 41%.
- Scheme 7.3: The synthetic strategy towards AHA and AHB. The preferred reaction conditions and product yields: i) HCOCOOH, MW, 110°C, 60 min, 54% (55) and 17 min, 60% (7); ii) H<sub>2</sub>, Pd/C, EtOAc, AcOH (~0.2%), 22 h, 41% (AHA) and 3 h, 47% (AHB).
- Scheme 7.4: The synthetic strategy towards GSH-Kyn, GSH-3OHKyn, Kyn yellow and 3OHKyn yellow from Kyn and 3OHKyn. The preferred reaction conditions and product yields: i) aqueous Na<sub>2</sub>CO<sub>3</sub>-NaHCO<sub>3</sub> (25 mM, pH 9.5), GSH, 37°C, 72 h, 51% (GSH-Kyn) and 44% (GSH-3OHKyn); ii) aqueous NaHCO<sub>3</sub> (0.49 M, pH ~9), reflux, 20 h, 26% (Kyn yellow) and 22% (3OHKyn yellow).
- Scheme 7.5: The synthetic strategy towards Cys-3OHKG from 3OHKG. The preferred reaction conditions: aqueous Na<sub>2</sub>CO<sub>3</sub>-NaHCO<sub>3</sub> (25 mM, pH 9.2), Cys, 37°C, 72 h, 35%. 240

#### LIST OF TABLES

117

Table 2.1: Glucosylation of methyl ester (10) with ABG.	47
Table 2.2: <sup>1</sup> H and <sup>13</sup> C NMR of 3OHKG and U-24. Spectra were obtained in $D_2O$ .	55
Table 4.1: Quantification of lens compounds in nuclei of normal and cataractous lenses (A	A) and

 Table 4.2: Extraction efficiency of UV filters and their metabolites from BLP. Analyses were performed in duplicate.
 120

cortices of normal lenses (B). Quantities are given in pmol/mg lens (dry mass).

- Table 5.1: Absorbance and fluorescence data for UV filter-treated BLP in 6 M guanidine hydrochloride(~2 mg/mL protein). N/D; not determined.162
- Table 5.2: Bound versus unbound UV filters following treatment of BLP with UV filters at pH 9.5 for 48 h. The percentage recovery was calculated from the initial amount of UV filters (*i.e.* AHA and AHB) or adducts (*i.e.* Cys-3OHKyn) found in the modified BLP upon acid hydrolysis (before guanidine hydrochloride extraction). GdHCl; guanidine hydrochloride. Quantification was conducted using the appropriate standard curves of Cys-3OHKyn, AHB and AHA. 179
- Table 5.3: Literature data on recovery of Kyn and 3OHKyn amino acid adducts after acid hydrolysis in the presence/absence of antioxidants and presence/absence of BLP. N/T, not tested; +, presence; -, absence.<sup>18,20,383</sup>
   180
- Table 5.4: Recovery of 3OHKG (3OHKyn), Cys-3OHKG (Cys-3OHKyn), AHB and AHA after acid hydrolysis for 24 h at 110°C in the presence of antioxidants and presence/absence of BLP. The experiment was done in duplicate.
  181
- Table 5.5:
   Absorbance and fluorescence data of dialysis buffer samples collected after protein modification at pH 9.5.

   188
- Table 5.6: Average concentrations of protein bound UV filters from in vitro and in vivo studies.
   190
- Table 6.1: Light intensities of the broad spectrum 125 W mercury arc lamp, both filtered and not<br/>filtered, were measured by a IL 1700 Research Radiometer.229

#### ABSTRACT

The kynurenine-based UV filters are unstable under physiological conditions and undergo side chain deamination, resulting in  $\alpha$ ,  $\beta$ -unsaturated carbonyl compounds. These compounds can react with free or protein bound nucleophiles in the lens via Michael addition. The key sites of the UV filters kynurenine (Kyn) and 3-hydroxykynurenine (30HKyn) modification in human lenses include cysteine (Cys), and to a lesser extent, lysine (Lys) and histidine (His) residues. Recent in vivo studies have revealed that 3-hydroxykynurenine-O-β-D-glucoside (30HKG) binds to Cys residues of lens crystallins in older normal human lenses. As a result of this binding, human lens proteins become progressively modified by UV filters in an agedependent manner, contributing to changes that occur with the development of age-related nuclear (ARN) cataract. Upon exposure to UV light, free UV filters are poor photosensitisers, however the role of protein-bound species is less clear. It has been recently demonstrated that Kyn, when bound to lens proteins, becomes more susceptible to photo-oxidation by UV light. Therefore, the investigation of 3OHKG binding to lens proteins, and the effect of UV light on proteins modified with 3OHKG and 3OHKyn, were major aims of this study. As a result of the role of these compounds as UV filters and their possible involvement in ARN cataract formation, it is crucial to understand the nature, concentration and modes of action of the UV filters and their metabolites present in the human lenses. Therefore, an additional aim was to investigate human lenses for the presence of novel kynurenine-based human lens metabolites and examine their reactivity.

As 3OHKG is not commercially available, to conduct protein binding studies, an initial aim of this study was to synthesise 3OHKG (Chapter 2). Through the expansion and optimisation of a literature procedure, 3OHKG was successfully synthesised using commercially available and inexpensive reagents, and applying green chemistry principles, where toxic and corrosive reagents were replaced with benign reagents and solvent-free and microwave chemistry was used. A detailed investigation of different reaction conditions was also conducted, resulting in either the improvement of reaction yields or reaction time compared to the literature method. Applying the same synthetic strategy, and using key precursors from the synthesis of 3OHKG, the UV filters 3OHKyn and  $4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid-<math>O-\beta$ -D-glucoside (AHBG), were also successfully synthesised (Chapter 3).

Chapter 4 describes the investigation of both normal and cataractous human lenses in an attempt to identify novel human lens metabolites derived from deaminated Kyn and 3OHKyn

(Chapter 4, Part A). Initially, 4-(2-aminophenyl)-4-oxobutanoic acid (AHA), glutathionyl-kynurenine (GSH-Kyn), kynurenine yellow (Kyn yellow), 4-(2-amino-3-hydroxyphenyl)-4-oxobutanoic acid (AHB), glutathionyl-3-hydroxykynurenine (GSH-3OHKyn) and 3-hydroxykynurenine yellow (3OHKyn yellow) were synthesised and human lenses were examined for their presence. AHA and AHB were synthesised from similar precursors to those used in the synthesis of 3OHKG, while the GSH adducts and yellow compounds were synthesised from Kyn and 3OHKyn *via* base induced deamination. Following isolation and structural elucidation, AHA, AHB and GSH-Kyn were confirmed as novel human lens metabolites. They were quantified in low pmol/mg lens (dry mass) levels in normal and cataractous lenses of all ages, while GSH-3OHKyn, Kyn yellow and 3OHKyn yellow were not detected. In contrast to AHA, the lens metabolites AHB, GSH-Kyn and GSH-SOHKyn were found to be unstable at physiological pH. The spectral properties of these compounds suggest that they may act as UV filters.

Chapter 4 (Part B) also describes the identification and characterisation of a novel human lens UV filter, cysteinyl-3-hydroxykynurenine-O- $\beta$ -D-glucoside (Cys-3OHKG). An authentic standard was synthesised *via* Michael addition of cysteine to deaminated 3OHKG. Cys-3OHKG was detected in low pmol/mg lens (dry mass) levels in normal lenses only after the 5<sup>th</sup> decade of life and was absent in cataractous lenses. Cys-3OHKG showed rapid decomposition at physiological pH.

Chapter 5 describes the identification and quantification of amino acids involved in covalent binding of 3OHKG to lens proteins. Model studies with bovine lens proteins and 3OHKG at pH 7.2 and 9.5 were undertaken. The amino acid adducts were identified *via* total synthesis and spectral analysis, and subsequently quantified upon acid hydrolysis of the modified lens proteins. Under both pH conditions, 3OHKG was found to react with lens proteins predominantly *via* Cys residues with low levels of binding also detected at Lys residues. Comparative studies with Kyn (pH 9.5) and 3OHKyn (pH 7.2 and 9.5) resulted in modified lens proteins at Cys residues, with only minor modification at Lys residues at pH 9.5. The extent of modification was found to be significantly higher at pH 9.5 in all cases. His adducts were not identified. 3OHKG-, Kyn- and 3OHKyn-modified lens proteins were found to be coloured and fluorescent, resembling those of aged and ARN cataractous lenses. In contrast, AHB and AHA, which can not form  $\alpha,\beta$ -unsaturated carbonyl compounds, resulted in noncovalent modification of lens proteins. AHB may contribute to lens colouration and fluorescence as further reactions of this material yielded species that have similar characteristics to those identified from 3OHKyn modification. These species are postulated to arise *via* auto-oxidation of the *o*-aminophenol moiety present in both 3OHKyn and AHB.

In Chapter 6, the potential roles of 3OHKG and 3OHKyn, and the related species AHA and AHB, in generating reactive oxygen species and protein damage following illumination with UV light was examined. The UV filter compounds were examined in both their free and protein-bound forms. Kyn-modified proteins were used as a positive control. Exposure of these compounds to UV light ( $\lambda$  305-385 nm) has been shown to generate H<sub>2</sub>O<sub>2</sub> and proteinbound peroxides in a time-dependent manner, with shorter wavelengths generating more peroxides. The yields of peroxides were observed to be highly dependent on the nature of the UV filter compound and whether these species were free or protein bound, with much higher levels being detected with the bound species. Thus, protein-bound 3OHKyn yielded higher levels of peroxide than 3OHKG, with these levels, in turn, higher than for the free UV filter compounds. AHB-treated lens proteins resulted in formation of low but statistically significant levels of peroxides, while AHA-treated lens proteins resulted in insignificant peroxide formation. The consequences of these photochemical reactions have been examined by quantifying protein-bound tyrosine oxidation products (3,4-dihydroxyphenylalanine [DOPA], di-tyrosine [di-Tyr]) and protein cross-linking. 3OHKG-modified proteins gave elevated levels of di-Tyr, but not DOPA, whereas 30HKyn-modified protein gave the inverse. DOPA formation was observed to be independent of illumination and most likely arose via oaminophenol auto-oxidation. AHB- and AHA-treated lens proteins resulted in statistically insignificant di-Tyr formation, while a light independent increase in DOPA was observed for both samples. Both reducible (disulfide) and non-reducible cross-links were detected in modified proteins following illumination. These linkages were present at lower levels in modified, but non-illuminated proteins, and absent from unmodified protein samples.

This work has provided an optimised synthetic procedure for 3OHKG and other lens metabolites (Chapters 2 and 3). Four novel lens metabolites have been identified and quantified in normal and cataractous human lenses (Chapter 4). Subsequent experiments, described in Chapter 5, identified the major covalent binding sites of 3OHKG to lens proteins, while AHA and AHB showed non-covalent binding. Further work described in Chapter 6 showed that protein-bound 3OHKG, Kyn and 3OHKyn were better photosensitisers of oxidative damage than in their unbound state. Together, this research has provided strong evidence that post-translational modifications of lens proteins by kynurenine-based metabolites and their interaction with UV light appear, at least in part, responsible for the age-

dependent colouration of human lenses and an elevated level of oxidative stress in older lenses. These processes may contribute to the progression of ARN cataract.

### THESIS DECLARATION

This thesis contains no material that has been accepted for the award of any higher degree or diploma at any University or Institution, and to the best of my knowledge contains no material previously published or written by another person, except where due reference is made in the text of the thesis.

#### ACKNOWLEDGMENTS

The production of this thesis involved many people other than the author and I wish to gratefully acknowledge the contribution of the following people:

My supervisors, A/Prof Joanne F. Jamie (Macquarie University, Sydney, NSW), Prof Michael J. Davies (The Heart Research Institute, Sydney, NSW) and Prof Roger J. W. Truscott (Save Sight Institute, Sydney, NSW), for their hard work, patience and professional support throughout the years, and from whom I have learnt a great deal.

Prof Peter Karuso for being my associate supervisor and for his expert advice on the NMR analysis, past NMR technical staff (Dr Andrew Piggott, Masoud Faroughi and Dr Isla Hains), MUCAB staff (Kieran Morgan) for assistance with the instrumentation, Prof Robert Willows and Prof Roger Hiller for useful discussions on fluorescence analysis, Dr Christopher McRae for maintaining our computer systems virus-free, Dr Ian Jamie for useful advice on statistical and fluorescence issues and Elsa Mardones for great laughs and maintaining the freeze dryer.

Dr Nicole Parker (formerly University of Wollongong, Wollongong, NSW) for the provision of Cys-Kyn and advice on protein extraction and UV light illumination experiments, and Dr Anastasia Korlimbinis (Save Sight Institute, Sydney, NSW) for providing Cys-3OHKyn and advice on acid hydrolysis of 3OHKyn-modified bovine lens proteins.

Dr Mano Abeysinghe and other members of Prof Roger Read's group (University of NSW, Sydney, NSW) for assistance with LC-MS.

Shimadzu Scientific Instruments staff, particularly Dr Paul Ahn, Sergio Zadro and Cedric Rebello for their kind help with LC-MS and HPLC.

Past and present members of Peter Karuso, Bridget Mabbutt, Andrew Try and Fei Liu's laboratories. It was such great fun to have you around and be stimulated by your endless energy to reach for more. Particularly, thank you Sudhir Shengule for useful discussions on NMR analysis and organic synthesis.

Linda MacDonald and Prof Gary Halliday (University of Sydney, Sydney, NSW) for the measurements of light intensity and spectral irradiance of the broad spectrum 125 W mercury

arc lamp, Dr Alexander Standler (Anton Paar, Graz, Austria) for expert advice on microwave application in organic synthesis, and John Allen (Australian National University, Canberra, ACT) for the HR-MS analyses.

Maria Hyland and Catherine Wong for administrative help through the years.

Dr Laurent Bornaghi for expert advice on sugar chemistry and great friendship.

All members of the Free Radical group at the Heart Research Institute (Sydney, NSW) for welcoming me to their laboratory and the great experience I gained during my stay. Particularly, my sincere thanks go to Dr Catherine Luxford and Dr Phillip Morgan for their expert assistance on the work discussed in Chapter 6.

Dr Peter Hains (Save Sight Institute, Sydney, NSW) for endless ES-MS/MS and HR-MS analyses, successful completion of work covered in Chapter 4 and reading sections of this thesis. It was great help to have you around and I am sure that this thesis would not progressed this way without your help. I can not say "thank you" enough.

My laboratory colleagues Dr Nynke Brouwer "the other half of the Yasnynke", Dr Christopher Austin and Dr Isla Hains for unmeasurable help and wonderful friendship, and thanks to everyone who has "lived" in office F7B 204: Boeddi, Azadeh, Anil, Olivier, Michelle, Chris, Gareth...

Dr Nynke Brouwer (Nynx), Joel Bedford (Joels) and Dr Nishen Naidoo (Mr Naidooo...) for great friendship and reading sections of this thesis.

My parents for their continuous love and support. Thank you mum Boja and dad Stevan, after all we have been through we are doing fine. Nobody can stop your dreams from coming true. Also, a big thanks to my brother Robert, sister-in-law Kristina and my lovely nephews Kristian (tetin miš) and Alek (tetina buba/mačak) for support throughout the years. Finally I would like to thank my boyfriend Damir Jaksic for his unconditional love, endless support and for believing in me during these years. Your assistance in preparation of this document is invaluable. Thank you! This thesis is dedicated to my mum and dad my brother, sister-in-law and nephews my boyfriend

Love you all forever and thank you

#### **PUBLICATIONS**

Sections of the work described in this thesis have been reported or are in the preparation for the following publications:

Peter G. Hains, Jasminka Mizdrak, Isla M. Streete, Joanne F. Jamie, Roger J. W. Truscott, Identification of the new UV filter compound cysteinyl-L-3-hydroxykynurenine-O- $\beta$ -D-glucoside in human lenses. *FEBS letters*, **2006**, 580, 5071-5076.

Jasminka Mizdrak, Peter G. Hains, Danuta Kalinowski, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie, Novel human lens metabolites from normal and cataractous human lenses. *Tetrahedron*, **2007**, 63, 4990-4999.

Jasminka Mizdrak, Peter G. Hains, Joanne F. Jamie, Michael J. Davies, Roger J. W. Truscott, Identification of lens protein-bound 3-hydroxykynurenine-*O*-β-D-glucoside. A possible role in Age-Related Nuclear cataract. *FEBS letters*, manuscript in preparation.

Jasminka Mizdrak, Peter G. Hains, Roger J. W. Truscott, Joanne F. Jamie, Michael J. Davies, Tryptophan-derived UV filter compounds covalently bound to lens proteins are photosensitizers of oxidative damage. *Free Radicals in Biology & Medicine*, manuscript in preparation.

Jasminka Mizdrak, Peter G. Hains, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie, A convenient synthesis of lenticular UV filters. *Tetrahedron Letters*, manuscript in preparation.

#### **CONFERENCE PRESENTATIONS**

The following oral or poster presentations have been made at conferences/symposia on research conducted during this PhD study:

- The ISMC/RACIOC Organic Chemistry Conference in Cairns (QLD, Australia), 4-8 July 2004. Jasminka Mizdrak, Danuta Kalinowski, Nicole R. Parker, Peter G. Hains, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie. Human lens chemistry -UV filters and cataract.
- The XVI International Congress of Eye Research in Sydney (NSW, Australia), 29 August-3 September 2004. Jasminka Mizdrak, Danuta Kalinowski, Nicole R. Parker, Peter G. Hains, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie. Human lens chemistry - UV filters and cataract.
- iii. The 12th Royal Australian Chemical Institute (RACI) Convention "Connect 2005" in Sydney (NSW, Australia), 3-7 July 2005. Jasminka Mizdrak, Peter G. Hains, Nicole R. Parker, Danuta Kalinowski, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie. Human Lens Chemistry - Age-Related Nuclear (ARN) cataract.
- iv. The Royal Australian Chemical Institute (RACI) Natural Products Group Annual one day symposium in Sydney (NSW, Australia), 30 September 2005. Jasminka Mizdrak, Peter G. Hains, Danuta Kalinowski, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie. Human Lens Natural Products.
- v. The 1st European Chemistry Congress in Budapest (Hungary), 28-31 August 2006. Jasminka Mizdrak, Peter G. Hains, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie. Human lens chemistry - UV filters and cataract. This international presentation resulted in media coverage, see Appendix C.
- Vi. One Day Royal Australian Chemical Institute (RACI) Organic Symposium in Canberra (ACT, Australia), 6 November 2006. Jasminka Mizdrak, Peter G. Hains, Roger J. W. Truscott, Michael J. Davies, Joanne F. Jamie. Human lens chemistry -Novel UV filters.

#### LIST OF ABBREVIATIONS

ABG 2,3,4,6-Tetra-O-acetyl- $\alpha$ -D-glucopyranosyl bromide AcOH Acetic acid Advanced glycation end AGE Arginine Arg Asn Asparagine Aspartic acid Asp 4-(2-Aminophenyl)-4-oxobutanoic acid AHA 4-(2-Amino-3-hydroxyphenyl)-4-oxobutanoic acid AHB AHBG 4-(2-Amino-3-hydroxyphenyl)-4-oxobutanoic acid-O-β-Dglucoside 4-(2-Amino-3-hydroxyphenyl)-4-oxobutanoic acid-O-AHBDG diglucoside Age-related nuclear ARN BAA β-Benzoylacrylic acid BLP Bovine lens proteins Boc Butyloxycarbonyl CH<sub>3</sub>CN Acetonitrile Cys-BAA Cysteinyl-β-benzoylbutanoic acid Cys-30HKG Cysteinyl-3-hydroxykynurenine-O-β-D-glucoside Dichloromethane DCM Dehydroascorbic acid DHA **Di-tyrosine** di-Tvr 3,4-Dihydrophenyalanine DOPA Deuterium oxide D<sub>2</sub>O ES-MS Electrospray mass spectrometry EtOH Ethanol Ethyl acetate EtOAc Emission Em Ex Excitation FOX Peroxide assay involving the oxidation of the Fe(II)-xylenol orange complex to the Fe(III) species Gln Glutamine Glycine Gly GSH Glutathione **GSH-3OHKG** Glutathionyl-3-hydroxykunurenine-*O*-β-D-glucoside **GSH-Kyn** Glutathionyl-kynurenine Glutathionyl-3-hydroxykynurenine GSH-3OHKyn Hydrochloric acid HCl Histidine His  $H_2O_2$ Hydrogen peroxide  $H_2SO_4$ Sulfuric acid Kynurenine Kyn Kynurenine yellow Kyn yellow LC-MS Liquid chromatography-mass spectrometry Lysine Lys MeOH Methanol Met Methionine MS/MS Tandem mass spectrometry

The following abbreviations are used throughout the text:

MW	Microwave
m/z	Mass to charge ratio
NAD(P)H	Reduced $\beta$ -nicotinamide dinucleotide (phosphate)
NMR	Nuclear magnetic resonance
δ	Chemical shift
NaOH	Sodium hydroxide
3OHKyn	3-Hydroxykynurenine
$^{1}O_{2}$	Singlet molecular oxygen in its ${}^{1}\Delta_{g}$ state
$O_2^{-1}$	Superoxide radical
30HKG	3 Hydroxykynurenine- $O$ - $\beta$ -D-glucoside
3OHKyn yellow	3-Hydroxykunurenine yellow
3OHKG yellow	3-Hydroxykunurenine-O-β-D-glucoside yellow
pet. spirit	Petroleum spirit
Phe	Phenylalanine
Pro	Proline
PSH	Protein sulfhydryl
PTM	Post-translational modification
RP-HPLC	Reversed phase-high performance liquid chromatography
RT	Room temperature
Rt	Retention time
SDS-PAGE	Sodium dodecyl sulphate-polyacrylamide gel electrophoresis
Ser	Serine
TFA	Trifluoroacetic acid
Thr	Threonine
TLC	Thin layer chromatography
Trp	Tryptophan
Tyr	Tyrosine
UDP	Uridine diphosphate
UV	Ultraviolet